

G2 Atomization Energies With Chemical Accuracy

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I. INTRODUCTION

II. THEORY

A. The DFT basis-set correction in a nutshell

The basis-set correction investigated here proposes to use the RSDFT formalism to capture a part of the short-range correlation effects missing in a finite one-electron basis-set. In a nutshell, this formalism relies on 1) the definition of a complementary density functional aiming at describing the correlation effects absent in a finite basis-set, 2) the definition of an *effective non divergent interaction* as the real-space representation of the coulomb operator projected in a finite basis-set, 3) the fit of such an effective interaction with a long-range interaction through the definition of a *range-separation parameter varying in space*, 4) the use of a correlation functional

from RSDFT with a *multi-determinant* reference evaluated with the range-separation parameter varying in space. More details can be found in².

1. Definition of basis-set dependent complementary functional

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III. RESULTS

A. The case of C₂ and the comparison with the F₁₂ methods.

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TABLE I. Dissociation energy (D_e) in kcal/mol of the C_2 molecule computed using FCIQMC, CIPSI, FCIQMC+F₁₂, CIPSI+LDA_{HF} and CIPSI+LDA_{HF-val} (valence only interaction and density) in the Dunning cc-pVXZ (VXZ) basis sets. ^a Results from Ref²

| | FCIQMC | CIPSI | FCIQMC+F ₁₂ | CIPSI+LDA _{HF} | CIPSI+LDA _{HF-val} | CIPSI+PBE _{HF} | CIPSI+PBE _{HF-val} |
|-----|--------|-------|------------------------------|-------------------------|-----------------------------|-------------------------|-----------------------------|
| V2Z | 130.0 | 132.0 | 142.3 | 141.9 | 142.9 | 142.7 | 145.6 |
| V3Z | 139.9 | 140.3 | 145.3 | 142.8 | 145.5 | 142.7 | 146.7 |
| V4Z | - | 143.6 | - | 145.8 | 146.2 | 145.3 | 147.0 |
| V5Z | - | 144.3 | - | 145.1 | 146.1 | 144.9 | 146.5 |
| | | | Estimated exact ^a | | | | |
| | | | 146.9 | | | | |

TABLE II. Dissociation energy (D_e) in kcal/mol of the N_2 molecule computed using FCIQMC, CIPSI, FCIQMC+F₁₂, CIPSI+LDA_{HF} and CIPSI+LDA_{HF-val} (valence only interaction and density) in the Dunning cc-pVXZ (VXZ) basis sets. ^a Results from Ref² taking into account the ZPE correction.

| | CIPSI | CIPSI+LDA _{HF} | CIPSI+LDA _{HF-val} | CIPSI+PBE _{HF} | CIPSI+PBE _{HF-val} |
|-----|-------|-------------------------|-----------------------------|-------------------------|-----------------------------|
| V2Z | 200.9 | 216.3 | 218.2 | 222.3 | 224.8 |
| V3Z | 217.1 | 223.1 | 225.8 | 224.6 | 226.7 |
| V4Z | 223.5 | 227.9 | 228.8 | 227.7 | 228.3 |
| V5Z | 225.7 | 227.9 | 228.4 | 227.7 | 228.3 |
| | | | Estimated exact | | |
| | | | 228.5 ^a | | |

TABLE III. Dissociation energy (D_e) in kcal/mol of the F_2 molecule computed using FCIQMC, CIPSI, FCIQMC+F₁₂, CIPSI+LDA_{HF} and CIPSI+LDA_{HF-val} (valence only interaction and density) in the Dunning cc-pVXZ (VXZ) basis sets. ^a Results from Ref² taking into account the ZPE correction.

| | CIPSI | CIPSI+LDA _{HF} | CIPSI+LDA _{HF-val} | CIPSI+PBE _{HF} | CIPSI+PBE _{HF-val} |
|-----|-------|-------------------------|-----------------------------|-------------------------|-----------------------------|
| V2Z | 27.5 | 30.8 | 31.1 | 32.1 | 32.3 |
| V3Z | 35.4 | 37.0 | 37.5 | 37.5 | 37.8 |
| V4Z | 37.5 | 38.7 | 38.8 | 38.7 | 38.8 |
| V5Z | | | | | |
| | | | Estimated exact | | |
| | | | 38.2 ^a | | |

TABLE IV. Dissociation energy (D_e) in kcal/mol of the O_2 molecule computed using FCIQMC, CIPSI, FCIQMC+F₁₂, CIPSI+LDA_{HF} and CIPSI+LDA_{HF-val} (valence only interaction and density) in the Dunning cc-pVXZ (VXZ) basis sets. ^a Results from Ref² taking into account the ZPE correction.

| | CIPSI | CIPSI+LDA _{HF} | CIPSI+LDA _{HF-val} | CIPSI+PBE _{HF} | CIPSI+PBE _{HF-val} |
|-----|-------|-------------------------|-----------------------------|-------------------------|-----------------------------|
| V2Z | 105.3 | 111.8 | 112.5 | 115.0 | 116.1 |
| V3Z | 114.6 | 117.2 | 118.5 | 118.4 | 119.4 |
| V4Z | 118.0 | 120.0 | 120.2 | 120.2 | 120.5 |
| V5Z | | | | | |
| | | | Estimated exact | | |
| | | | 120.2 ^a | | |